Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.009 Å R factor = 0.056 wR factor = 0.155 Data-to-parameter ratio = 8.7

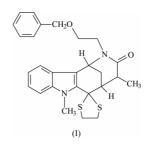
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved The title compound, $C_{27}H_{30}N_2O_2S_2$, consists of a pentacyclic ring system containing a carbazolenine skeleton with an azocino group, two methyl groups and a dithiolane ring as substituents.

Received 12 November 2003 Accepted 24 November 2003 Online 6 December 2003

Comment

1,2,3,4-Tetrahydrocarbazole derivatives can be considered to be synthetic precursors of tetracyclic indole alkaloids, having tricyclic ring systems such as those in the *Strychnos* family of indole alkaloids (Bosch & Bonjoch, 1988). Tetrahydrocarbazole systems are present in the framework of a number of indole-type alkaloids of biological interest (Phillipson & Zenk, 1980; Saxton, 1983; Abraham, 1975).



The structures of tricyclic and pentacyclic ring systems, with dithiolane and other substituents of the tetrahydrocarbazole core, have been the subject of much interest in our laboratory. These include 1,2,3,4-tetrahydrocarbazole-1-spiro-2'-[1,3]dithiolane, (II) (Hökelek et al., 1994), N-(2-methoxyethyl)-N-{2,3,4,9-tetrahydrospiro[1H-carbazole-1,2-(1,3)dithiolane]-4yl}benzenesulfonamide, (III) (Patır et al., 1997), spiro[carbazole-1(2H),2'-[1,3]-dithiolan]-4(3H)-one, (IV) (Hökelek et al., 1998), 9-acetonyl-3-ethylidene-1,2,3,4-tetrahydrospiro-[carbazole-1,2'-[1,3]dithiolan]-4-one, (V) (Hökelek et al., 1999), and N-(2,2-dimethoxyethyl)-N-{9-methoxymethyl-1,2,3,4-tetrahydro-spiro[carbazole-1,2'-[1,3]dithiolan]-4-yl}benzamide, (VI) (Hökelek & Patır, 1999); also the pentacyclic 6-ethyl-4-(2-methoxyethyl)-2,6-methano-5-oxocompound hexahydropyrrolo(2,3-d)carbazole-1-spiro-2'-(1,3)dithiolane, (VII) (Hökelek & Patır, 2002).

The hexahydro-1,5-methano[4,3-*b*]indole basic skeletal framework can be considered to be a synthetic precursor of pentacyclic indole alkaloids. These alkaloids share the *ABCDE* ring system as a common structural element, and include a large group of naturally occurring compounds, with highly complex structures (Rahman & Basha, 1983). They have pharmacological and extremely toxic properties.

Compound (I) (Fig. 1) consists of a pentacyclic system containing a carbazolenine skeleton with an azocino, two methyl groups and a dithiolane ring as substituents at posi-

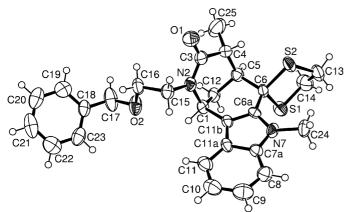


Figure 1

An *ORTEP*-3 (Farrugia, 1997) drawing of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

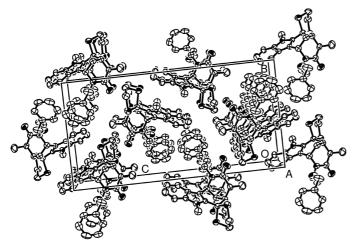


Figure 2

A packing diagram, viewed down the *a* axis. H atoms have been omitted.

tions N2, N7, C4 and C6, respectively. As can be seen from the packing diagram (Fig. 2), the molecules are stacked along the a axis. Dipole–dipole and van der Waals interactions are effective in the molecular packing.

The S atoms of the dithiolane ring have electron-releasing properties, but the N atom at position 7 and the O atom attached to C3 have electron-withdrawing properties, leading to some changes in the bond lengths and angles of the carbazolenine skeleton.

Some angles in the carbazolenine skeleton are compared with those in the structures of other tricyclic ring systems containing the dithiolane substituent (Table 2). The bond lengths N7–C7a and N7–C6a agree well with those in compounds (V) and (VI) [1.370 (8), 1.377, 1.376; 1.392 (8), 1.396 and 1.391 Å, respectively]. In all three structures atom N7 is substituted.

An examination of the deviations from the least-squares planes through individual rings shows that rings *A* (C7a/C8/C9/C10/C11/C11a), *B* (N7/C7a/C11a/C11b/C6) and *F* (C18–C23) are planar. Rings *C* (C1/C11b/C6a/C6/C5/C12), *D* (C1/N2/C3/C4/C5/C12) and *E* (C6/S1/C14/C13/S2) are, of course, not planar.

The structure reveals a number of short contacts $S1 \cdots H5(C5) = 2.777(2)$, $S2 \cdots H4(C4) = 2.483(2)$, $O1 \cdots H15A(C15) = 2.379(5)$, $O1 \cdots H25B(C25) = 2.585(5)$, $O1^{i} \cdots H13A(C13) = 2.486(5)$ and $O2 \cdots H1(C1) = 2.651(5)$ Å [symmetry code: (i) -x, $y - \frac{1}{2}, \frac{1}{2} - z$].

Experimental

Crystal data

The title compound, (I), was prepared from sodium hydride (24.0 mg, 1.0 mmol) and *N*-(2-benzoyloxyethyl)-6-(1,3-dithiolan-2-yl)-1,2,3,4,-5,6-hexahydro-1,5-methano-2-azocino[4,3-*b*]indol-2-one (450.0 mg, 1.0 mmol) in THF (30 ml) and methyl iodide (1.0 ml). The mixture was heated at reflux for 4 h under a nitrogen atmosphere. The mixture was later cooled in an ice bath and methanol (5 ml) and water (20 ml) were added. After extraction with chloroform (30 ml), the organic layer was dried with MgSO₄ and the solvent was evaporated. The residue was crystallized from benzene:*n*-hexane (1:9) (yield 390.0 mg, 84%), m.p. 418.6 K.

$\begin{split} M_r &= 478.65 \\ \text{Orthorhombic, } P2_12_12 \\ a &= 7.8787 (10) \text{ Å} \\ b &= 12.446 (2) \text{ Å} \\ c &= 24.501 (3) \text{ Å} \\ V &= 2402.5 (6) \text{ Å}^3 \\ Z &= 4 \\ D_x &= 1.323 \text{ Mg m}^{-3} \end{split}$	1	Mo K α radiation Cell parameters from 25 reflections $\theta = 10-18^{\circ}$ $\mu = 0.25 \text{ mm}^{-1}$ T = 293 (2) K Block, colourless $0.35 \times 0.25 \times 0.15 \text{ mm}$			
Data collection					
Enraf–Nonius CAD-4 diffractometer Non-profiled ω scans Absorption correction 2790 measured reflection 2790 independent reflections with <i>I</i>	ions ections	$\theta_{\text{max}} = 26.3^{\circ}$ $h = 0 \rightarrow 9$ $k = 0 \rightarrow 15$ $l = -30 \rightarrow 0$ 3 standard reflections frequency: 120 min intensity decay: 1%			
Refinement					
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.056$ $wR(F^2) = 0.155$ S = 1.03 2543 reflections 292 parameters Table 1		H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0741P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.28 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{\text{min}} = -0.28 \text{ e} \text{ Å}^{-3}$			
2543 reflections 292 parameters Table 1		$\Delta \rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$			
2543 reflections 292 parameters	arameters (Å,	$\Delta \rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$			
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6	1.843 (7)	$\Delta \rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$ °). C11a-C11	1.412 (7)		
2543 reflections 292 parameters Table 1 Selected geometric p 51–C6 S2–C6	1.843 (7) 1.867 (6)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $^{\circ}).$ $C11a-C11$ $C11a-C7a$	1.421 (7)		
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6 S2-C6 N2-C3	1.843 (7) 1.867 (6) 1.339 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}^{\circ}).$ $\boxed{\begin{array}{c} C11a - C11\\ C11a - C7a\\ C7a - C8\end{array}}$	1.421 (7) 1.390 (6)		
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6 S2-C6 N2-C3 N7-C7a	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $^{\circ}).$ $\hline \begin{array}{c} \text{C11a-C11} \\ \text{C11a-C7a} \\ \text{C7a-C8} \\ \text{C5-C6} \end{array}$	1.421 (7) 1.390 (6) 1.561 (7)		
2543 reflections 292 parameters Table 1 Selected geometric p S1–C6 S2–C6 N2–C3 N7–C7a N7–C6a	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}^{\circ}).$ $\boxed{\begin{array}{c} C11a - C11\\C11a - C7a\\C7a - C8\\C5 - C6\\C8 - C9\end{array}}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8)		
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6 S2-C6 N2-C3 N7-C7a N7-C6a	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $^{\circ}).$ $\hline \begin{array}{c} \text{C11a-C11} \\ \text{C11a-C7a} \\ \text{C7a-C8} \\ \text{C5-C6} \end{array}$	1.421 (7) 1.390 (6) 1.561 (7)		
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6 S2-C6 N2-C3 N7-C7a N7-C7a N7-C6a C6a-C6 C11a-C11b	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.497 (8) 1.401 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$). C11a-C11 $C11a-C7a$ $C7a-C8$ $C5-C6$ $C8-C9$ $C11-C10$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8)		
2543 reflections 292 parameters Table 1 Selected geometric p 51–C6 52–C6 N2–C3 N7–C7a N7–C7a N7–C6a C6a–C6 C11a–C11b C3–N2–C1	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7)		
2543 reflections 292 parameters 292 parameters 292 parameters S1–C6 52–C6 52–C6 N2–C3 N7–C7a N7–C7a N7–C6a C6a–C6 C11a–C11b C3–N2–C1 O1–C3–N2	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6) 121.0 (7)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}^{\circ}).$ $\begin{array}{c} C11a - C11\\C11a - C7a\\C7a - C8\\C5 - C6\\C8 - C9\\C11 - C10\\C9 - C10\\C9 - C10\\C16 - O2 - C17\\N7 - C7a - C8\end{array}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6) 129.7 (6)		
2543 reflections 292 parameters Table 1 Selected geometric p S1-C6 S2-C6 N2-C3 N7-C7a N7-C6a C6a-C6 C11a-C11b C3-N2-C1 O1-C3-N2 C11b-C6a-N7	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6)		
2543 reflections 292 parameters Table 1 Selected geometric p 51-C6 52-C6 N2-C3 N7-C7a N7-C6a C6a-C6 C11a-C11b C3-N2-C1 O1-C3-N2 C11b-C6a-N7 C11b-C6a-C6	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6) 121.0 (7) 107.9 (5)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6) 129.7 (6) 107.4 (6)		
2543 reflections 292 parameters Table 1 Selected geometric p 51-C6 52-C6 N2-C3 N7-C7a N7-C6a C6a-C6 C11a-C11b C3-N2-C1 O1-C3-N2 C11b-C6a-N7 C11b-C6a-C6 N7-C6a-C6	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6) 121.0 (7) 107.9 (5) 124.6 (6)	$\Delta \rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}$). $\begin{array}{c} \text{C11a-C11} \\ \text{C11a-C7a} \\ \text{C7a-C8} \\ \text{C5-C6} \\ \text{C8-C9} \\ \text{C11-C10} \\ \text{C9-C10} \\ \hline \end{array}$ $\begin{array}{c} \text{C16-O2-C17} \\ \text{N7-C7a-C8} \\ \text{N7-C7a-C8} \\ \text{N7-C7a-C11a} \\ \text{C8-C7a-C11a} \\ \hline \end{array}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6) 129.7 (6) 107.4 (6) 122.8 (7)		
2543 reflections 292 parameters Table 1 Selected geometric p 51-C6 82-C6 N2-C3 N7-C7a N7-C6a C6a-C6 C11a-C11b C3-N2-C1 O1-C3-N2 C11b-C6a-N7 C11b-C6a-C6	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6) 121.0 (7) 107.9 (5) 124.6 (6) 127.2 (5)	$\Delta \rho_{\text{max}} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}$). C11a-C11 C11a-C7a C7a-C8 C5-C6 C8-C9 C11-C10 C9-C10 C16-O2-C17 N7-C7a-C8 N7-C7a-C8 N7-C7a-C1a C8-C7a-C11a C6a-C11b-C11a	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6) 129.7 (6) 107.4 (6) 122.8 (7) 109.0 (6)		
2543 reflections 292 parameters Table 1 Selected geometric p 51-C6 52-C6 N2-C3 N7-C7a N7-C6a C6a-C6 C11a-C11b C3-N2-C1 O1-C3-N2 C11b-C6a-N7 C11b-C6a-C6 N7-C6a-C6 C11b-C1a-C11	1.843 (7) 1.867 (6) 1.339 (8) 1.370 (8) 1.392 (8) 1.487 (8) 1.401 (8) 122.5 (6) 121.0 (7) 107.9 (5) 124.6 (6) 127.2 (5) 136.0 (7)	$\Delta \rho_{max} = 0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{min} = -0.28 \text{ e } \text{\AA}^{-3}$ $\overset{\circ}{}^{\circ}).$ $\begin{array}{c} C11a-C11\\C11a-C7a\\C7a-C8\\C5-C6\\C8-C9\\C11-C10\\C9-C10\\\end{array}$ $\begin{array}{c} C16-O2-C17\\N7-C7a-C8\\N7-C7a-C11a\\C8-C7a-C11a\\C8-C7a-C11a\\C6a-C11b-C11a\\C6a-C11b-C1\\\end{array}$	1.421 (7) 1.390 (6) 1.561 (7) 1.359 (8) 1.380 (8) 1.413 (7) 111.9 (6) 129.7 (6) 107.4 (6) 122.8 (7) 109.0 (6) 121.8 (6)		

Table 2

Comparison of the bond angles ($^{\circ}$) in the carbazole core of (I) with the corresponding values in the related compounds (II), (III), (IV), (V) and (VI).

	(I)	(II)	(III)	(IV)	(V)	(VI)
C6-C6a-N7	127.2 (5)	125.0 (3)	124.1 (7)	126.4 (2)	127.5 (2)	126.7 (2)
C11b-C11a-C11	136.0 (7)	133.6 (4)	136.3 (8)	134.7 (2)	134.0 (3)	134.7 (2)
N7-C7a-C8	129.7 (6)	130.8 (4)	128.1 (9)	129.8 (2)	129.4 (3)	129.1 (2)
S1-C6-C6a	106.6 (4)	110.2 (3)	112.1 (5)	112.8 (1)	115.9 (1)	115.9 (2)
C6a-N7-C7a	109.2 (5)	108.6 (3)	108.1 (6)	109.6 (1)	108.1 (1)	108.1 (2)

Most of the reflections were weak probably due to the crystal quality. H atoms were positioned geometrically and refined with a riding model, with C—H in the range 0.93–0.98 Å. $U_{\rm iso}$ values were constrained to be 1.2 (1.3 for methyl groups) times $U_{\rm eq}$ of the carrier atom. The Flack (1983) parameter refined to a value of 0.7 (2), making the absolute configuration indeterminate.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The authors acknowledge the purchase of the CAD-4 diffractometer under grant DPT/TBAG1 of the Scientific and Technical Research Council of Turkey.

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